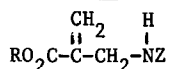


ALPHA-OXYGENATED FATTY ACIDS OCCURRING
AS AMIDES OF 2-METHYLENE- β -ALANINE IN A MARINE SPONGE

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In 1973 Kashman *et al.*² reported the isolation of a new amino acid 2-methylene- β -alanine (1a) from the Red Sea sponge *Fasciospongia cavernosa*. It occurs as the methyl ester (1b) and its amino group is acylated by five or more C₁₆-C₂₀ fatty acids (1c, n = 13-17). We have encountered



<u>1</u>	<u>a</u>	R = Z = H	<u>b</u>	R = Me, Z = H
	<u>c</u>	R = Me, Z = OC-CH ₂ -(CH ₂) _n -CH ₃		
	<u>d</u>	R = Me, Z = OC-CO-(CH ₂) _n -CH ₃		
	<u>e</u>	R = Me, Z = OC-CHOH-(CH ₂) _n -CH ₃		
	<u>f</u>	R = H, Z = OC-CH ₂ -(CH ₂) _n -CH ₃		

four sets of N-acyl derivatives of the same amino acid in an unidentified black sponge from Hawaiian waters.³ In our case, the amino acid occurs free or as the methyl ester. More remarkably, the amide linkage in two groups of compounds is formed with three C₁₄-C₁₆ fatty acids that bear unprecedented alpha keto or alpha hydroxy functions.

Continuous pentane extraction of the freeze dried sponge followed by silica gel and Sephadex chromatography of the pentane residue yielded the major group of compounds, 1d (n = 11, 12, 13), as an amorphous white wax, mp 36.5-37°. High resolution mass spectrometry showed this solid to be a mixture of three homologs predominantly, C₁₉H₃₃NO₄, C₂₀H₃₅NO₄, and C₂₁H₃₇NO₄, differing only in length of the aliphatic chain. The ir spectrum indicated presence of a secondary amide (3400 sh, 3420, 1665, 1520 and 1260), an ester (1720), and a terminal methylene (950 cm⁻¹). We made no attempt to separate this or the analogous (*vide infra*) amide mixtures. Comparison of our ¹H nmr data with those of Kashman *et al.*² (Table 1) demonstrated close correspondence of the 2-methylene-propanoic acid moiety, but substantial downfield shifts of the amide proton and of the methylene protons alpha to the acyl function.

¹³C nmr data supported the 2-methylene- β -alanine (C-1 166.0, C-2 135.5, C-3 40.2, C-4 127.4) as well as the amide function (C-1' 159.8 ppm), but also revealed an additional carbonyl function (C-2') resonating at 198.7 ppm. The ¹H chemical shift data (Table 1) demand that this carbonyl be placed next to the amide carbonyl. This unusual structural feature is further substantiated by a uv maximum at 244 nm (ϵ 2900, dioxane), in reasonable agreement with a literature value of 241 nm (ϵ 2456, EtOH).⁴ Mass spectral fragmentation is in full accord with the postulated structure. The three main parent peaks at m/e 367, 353, and 339 undergo facile cleavage between the

Table 1. Proton Chemical Shift Data (δ) for Our Compounds 1d ($n = 11-13$) and Kashman's 1c ($n = 13-17$); $Y = (CH_2)_nCH_3$

		1	2		3		1'	2'	3'	
	Me -	O ₂ C -	C	-----	CH ₂ -	N	-----	CO -	X -	CH ₂ Y
			HCH			H				
			4							
Our <u>1d</u> ,		3.74	6.25	5.77	4.09	7.35	--	--	2.85	
X = CO										
Kashman's <u>1c</u> ,		3.74	6.22	5.79	4.06	6.12	--	2.17	--	
X = CH ₂										

carbonyls, leading to the base peak at m/e 142 ($C_6H_8NO_3$) and acyl fragments at m/e 225 ($C_{15}H_{29}O$) and corresponding lower homologs at m/e 211 and 197.

The polar fractions of the original silica gel chromatography were rechromatographed on Sephadex to remove high molecular weight material, and again on silica gel, which yielded the second most abundant group of compounds, 1e, ($n = 11-13$) and a mixture of 1c and 1f. This mixture was separable on Sephadex LH-20. The 1c amides were identical by 1H and tlc comparison as well as by ms -- except of course for chain length -- with a sample provided by Professor Kashman. The 1H nmr spectrum of the 1f amides was virtually superimposable on that of 1c except for the absence of the methyl ester and the presence of a D_2O exchangeable proton signal at δ 10.12. Mass spectral parent peaks occurred at 14 mass units below those of 1c, as did the base peak, now at m/e 128. Esterification of 1f with ethereal diazomethane yielded a group of compounds identical with 1c by tlc, ms, and 1H nmr. The attempted reverse reaction, mild acid or base hydrolysis of 1c, failed and led only to decomposition of the β -alanine portion of the molecule.

The 1e amides were characterized by a broad ir band at 3300 cm^{-1} , an exchangeable proton in the 1H nmr spectrum at about δ 3.7, and a 3H signal at δ 4.07 assigned to the C-3 methylene alpha to the NH (dd, $J_{3NH} = 6\text{ Hz}$, $J_{34} < 1\text{ Hz}$) and to a secondary alcohol methine (m). The 1H nmr spectrum further showed the absence of a 2H triplet between δ 2 and 3, characteristic of a methylene next to a carbonyl. The likely surmise that 1e is the alcohol corresponding to 1c was proven by oxidizing the hydroxyamides 1e with chromium trioxide dipyridine complex to a group of ketoamides identical with 1c by tlc, 1H nmr, and ms.

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2. Y. Kashman, L. Fishelson, and I. Neeman, Tetrahedron **29**, 3655 (1973).
3. The sponge was collected near the Blowhole, Oahu. It is purple to black on the outside and off-white inside. Pending identification we are keeping specimens and photographs on file.
4. E. W. Colvin, J. Martin, W. Parker, R. A. Raphael, B. Shroot, and M. Doyle, J. Chem. Soc. Perkin Trans I, 860 (1972).

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